Reg. No.						



## CH/AC/OC/CA 402

## First Semester M.Sc. Degree Examination, December 2018 Chemistry/Applied Chemistry/Organic Chemistry/Analytical Chemistry (CBCS Scheme) (2015 Batch) (Repeaters) ORGANIC CHEMISTRY

Time: 3 Hours Max. Marks: 70

**Note**: 1) Answer **any ten** sub-divisions from Part **A** and **any five** questions from Part **B**.

2) Figures to the **right** indicate marks.

1. Answer any ten of the following:

 $(10 \times 2 = 20)$ 

a) Predict the product for the following with justification.

RCOO
$$^{t}$$
Bu +  $H_{2}$ O $^{18}$  +  $H^{+}$  -------?

- b) With suitable example differentiate localized and delocalized bonding.
- c) Among the following which one is more basic and why? Aniline and methyl amine.
- d) Among the following which one is aromatic and why?







- e) Write the conformational structure of cellulose.
- f) Predict the products for the following reactions. Justify your answer.

- g) Write the cis-and trans-forms of 1, 4 cyclohexane dicarboxylic acid.
- h) With suitable example differentiate stereoselective and stereospecific reactions.
- i) Write the stable conformational structure for  $\alpha$ -D-glucopyranose and  $\alpha$ -L-gulopyranose.



- j) Starch gives blue colouration with iodine but cellulose doesnot. Justify this statement.
- k) Name three rearrangements which involves nitrene as intermediates.
- I) Mention the non-kinetic methods available for investigating reaction mechanism.

## PART - B

Answer any five questions:

 $(5 \times 10 = 50)$ 

- 2. a) With suitable example, distinguish between inductive effect and mesomeric effect.
  - b) With suitable examples, explain the effect of solvent on base strength.
  - c) Discuss the generation and reactions of carbene.

(4+3+3)

- 3. a) With suitable example, distinguish between resonance and tautomerism.
  - b) Write a note on (i) fluxanal molecules and (ii) catenanes.
  - c) With examples, explain Bronsted-Lowry acid-base concept. (4+3+3)
- 4. a) How isotope labeling will help to reduce the possible mechanism for benzyne mechanism? Explain with suitable example.
  - b) Addition of bromine across the double bond always leads to trans-product.

    Justify this statement with suitable example. (5+5)
- 5. a) Write the most stable conformational structures for cis and trans 1, 2-dimethyl cyclohexane and decalin.
  - b) With justification, predict the product for the following:

$$\begin{array}{cccc}
\text{HO} & \text{O} \\
\text{Ph} & \text{H}
\end{array}$$

c) Give the R or S notations for the following.



- 6. a) With suitable example, explain the effect of conformation on reactivity.
  - b) Draw the Fischer projection formula of each stereoisomer of 2, 3 dibromobutane. Indicate the isomeric relationship between each pair of stereoisomer.
  - c) With suitable examples explain any one method for the resolution of racemic mixture. (4+3+3)
- 7. a) What is muta rotation? Explain the mechanism with suitable example.
  - b) Cellobiose has  $\beta(1\rightarrow 4)$  linkage. Justify.
  - c) Explain the role of cuprammonium solution in establishing the conformation of monosaccharides. (4+3+3)
- 8. a) Using periodic oxidation method, how do you prove that cellulose has  $1-\rightarrow 4$  linkage.
  - b) Explain the different methods used for linkage analysis of polysaccharides.
  - c) Write a note on chitin. (4+3+3)
- 9. Explain with suitable examples how the following techniques are useful in the determination of reaction mechanism.
  - i) Stereochemical evidences
  - ii) Trapping of reaction intermediates

iii) Limitation of an reaction.	(4+3+3)
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